

Emulsomes : A Novel Vesicular Platform for Enhanced Drug Delivery and Targeted Therapy

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Abstract

Over the past several decades, there has been a significant increase in interest in the development of innovative drug delivery systems (NDDS), focusing on creating systems that administer drugs at therapeutic rates and target specific action sites. Conventional drug forms, even those with extended release, cannot meet these requirements, while NDDS aims to optimize therapeutic effects and minimize side effects. Emulsomes, a type of NDDS, are lipid-based vesicular systems combining the properties of liposomes and emulsions. They are made up of a phospholipid bilayer enclosing a solid or semi-solid lipid core, enabling the controlled release and encapsulation of both lipophilic and hydrophilic medications. Emulsomes offer stability advantages over traditional liposomes, making them suitable for delivering drugs with a poor water solubility, such as antifungal and anticancer agents, while reducing toxicity and improving bioavailability. Various preparation methods, such as lipid film formation, reverse phase evaporation, and sonication, contribute to the formation of emulsomes with specific characteristics. The flexibility in emulsome composition allows for tailored applications in drug delivery, enhancing the targeted and sustained release of therapeutic agents.

Keywords: Emulsomes, Novel Drug Delivery Systems, Lipid-Based Vesicles, Controlled Release, Targeted Delivery, Drug Encapsulation.

Introduction

The development of novel drug delivery systems (NDDS) has drawn a lot of interest in recent decades. Two requirements should ideally be met by the NDDS: first, it should route the active component to the site of action; second, it should provide the drug at a pace chosen by the body during therapy. Not even normal extended-release dose formulations can satisfy all of requirements. Novel drug delivery aims to minimize unwanted side effects while maintaining a reasonably constant, effective drug level in the body, or to continue drug activity at a predefined pace. Another way to improve drug efficacy is to place controlled release devices in or close to the organ or tissue that is impacted.

As an alternative, it can target drug action by delivering the medication to certain cell types through the use of carriers or chemical changes.[1] Pharmaceutical carriers come in a variety of forms, including particulate, polymeric, macromolecular, and cellular. Colloidal carrier systems, also known as particulate type carriers, include vesicles, lipid particles, microspheres, nanoparticles, and polymeric micelles.[2] These are colloidal particles called vesicles, in which an aqueous compartment is encircled by an amphiphilic molecule-based concentric bilayer. This efficient vehicle can be used to transport hydrophilic drugs, which are found in the inner aqueous compartment, as well as hydrophobic drugs, which attach to the lipid bilayer. The vesicular

systems like liposomes, emulsomes, sphingosome, niosomes, transferosomes, pharmacosomes, virosomes, ethosomes, aquasomes, bilosomes etc.

Emulsomes

Emulsome is a lipid-based drug delivery system specifically developed to enhance the parenteral delivery of drugs with low water solubility. [3,4] Emulsomes are a unique type of lipid vesicle, featuring an outer phospholipid bilayer and a solid fat core within.(Figure 1). The inner core of emulsomes consists of fats and triglycerides, stabilized as an oil-in-water emulsion through the use of high concentrations of lecithin. The characteristics of emulsions and liposomes are combined in emulsomes. Their solidified or semi-solidified oily core allows for a high concentration of lipophilic drugs and facilitates controlled release. In addition, emulsomes can encapsulate water-soluble drugs within the aqueous compartments formed by the surrounding phospholipid layers.

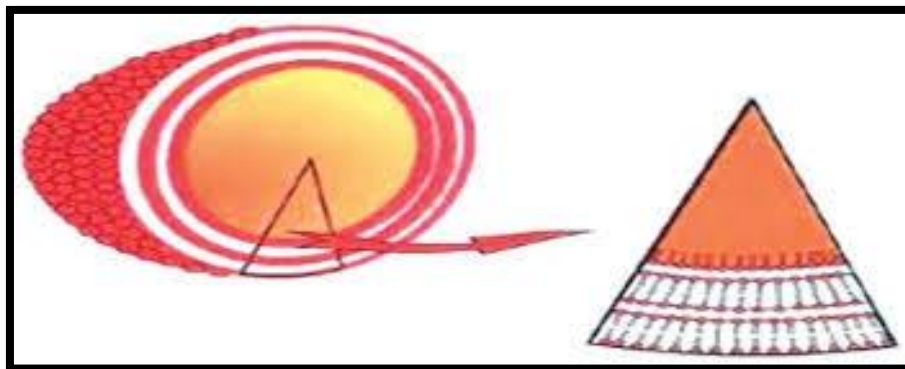


Figure.1 : Emulsome

Emulsomes, free from solvents and surfactants, have demonstrated an impressive capacity to encapsulate water-insoluble antifungal and anticancer drugs, resulting in improved drug delivery and enhanced preclinical efficacy, particularly in oral administration.[5] Emulsomes offer greater stability compared to liposomes, which are often less stable. By adopting emulsomes, you can avoid common problems like oxidation, sensitivity to hydrolysis, and aggregation that are linked to conventional liposomes and other vesicular delivery systems. By encapsulating the active ingredient within a vesicular structure, emulsomes help extend the drug's circulation time in the body and reduce toxicity. Traditional drug delivery methods are often not suited to meet these needs.[6] Since these medications are lipophilic (poorly soluble in water), they cannot be taken or, if they were, would have unfavorable side effects. For this reason, emulsomes are very helpful in this regard. Encapsulating a wide range of medications within emulsomes, such as antifungal, neuroprotective, AZT derivatives, β -blockers, antiepileptic, antibiotic, antineoplastic, and anti-inflammatory agents, is possible.[7] Emulsomes make lipophilic medications more soluble and bioavailable, and their structure permits controlled or prolonged drug release. After loading, the drug is sonicated to produce microscopic emulsomes.[8,9] Emulsomes' positive charge makes it expected that they will impede lysosomal breakdown and guarantee that the medication is taken internally.[9] Research has shown that emulsomes effectively transport a variety of protein and peptide drugs, including curcumin, amphotericin B (AmB), azidothymidine (AZT), methotrexate (MTX), and dithranol.[9,10,11,]

A researcher prepared emulsomes containing amphotericin B, a potent antifungal and anti-leishmanial drug, were used to treat visceral leishmaniasis. This study aims to develop macrophage-targeted emulsomes, specifically for the liver, spleen, and bone marrow, to reduce the adverse effects associated with conventional treatments.[12] For the treatment of psoriasis, Raza et al. created dithranol-loaded emulsomes that exhibit enhanced stability, effectiveness, and biocompatibility.[13] According to Kamboj et al., Vyas et al. created zidovudine-loaded emulsomes with the goal of treating serious viral illnesses such Epstein-Barr virus, HIV, and hepatitis. These emulsomes are intended for targeted and sustained delivery to the liver.[14,4].

Structure of emulsome

Emulsomes are lipid-based nanoparticles commonly used for drug delivery. their composition, consisting of an oil phase encased in a phospholipid bilayer, can be likened to a hybrid between liposomes and a solid lipid nanoparticles. Emulsomes contain phospholipid bilayers arranged head to tail, with the hydrophilic heads of the two layers facing outward and the hydrophobic tails facing inward. This bilayer functions to prevent the encapsulated oil phase from degrading and provides the emulsomes with a stable structure.[15,16] emulsomes can contain more than just the phospholipid bilayer and oil phase; they can also contain surfactants, stabilizers, and targeting ligands, among other materials.[17,18] These additional chemicals could help improve the emulsomes' ability to target specific tissues or cells, as well as their efficacy and longevity.[19] The general structure of emulsomes is highly flexible, allowing it to be tailored to the specific needs of different drug delivery applications. By carefully selecting the ingredients and refining the formulation, It is feasible to develop emulsomes that are not only effective and stable but also well-tolerated by the body.[20,21] (figure-2)

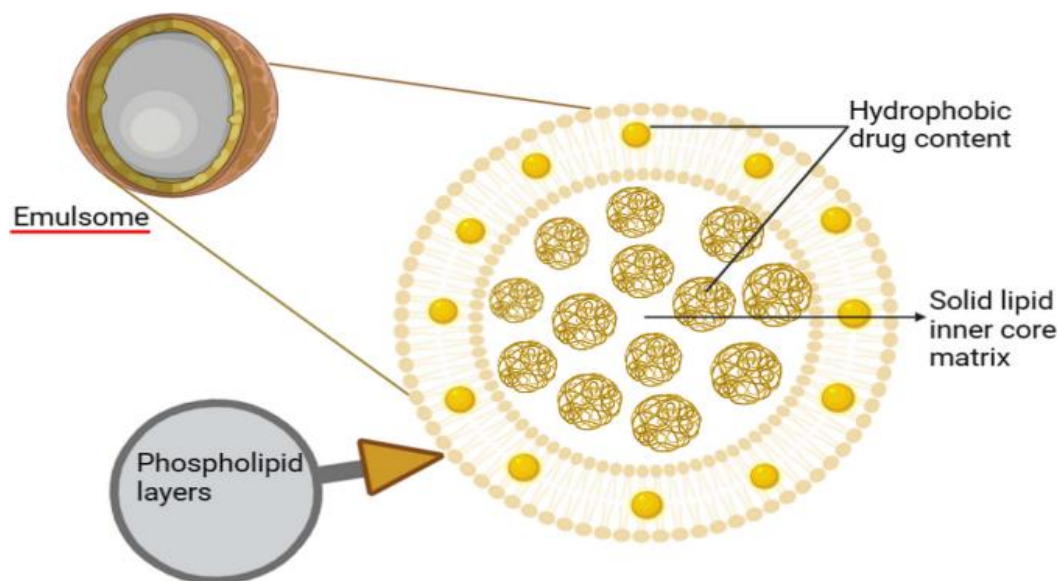


Figure 2 : structure of emulsome

Components or Formulation of Emulsomes –

Antioxidant

The lipid core of the emulsome particles in this invention may optionally contain one or more antioxidants. As members of the vitamin E family, α -tocopherol or its derivatives are the recommended antioxidants. Butylated hydroxytoluene is one of the other antioxidants (BHT). Antioxidants reduce the production of peroxides, which are results of unsaturated lipid oxidative breakdown. Using saturated fatty acids to prepare the lipid core can help reduce the need for antioxidants.[22]

Lipid core

The internal hydrophobic core, or lipid core, is a key component of emulsomes. It is made up of lipids and can exist in solid, lipid, or a mixed solid and liquid crystal phase at room temperature (25°C). There are many lipids and lipid-like excipients available on the market. In medical terminology, they are collectively referred to as lipids.[23] A single lipid or a blend of lipids can be used. These are molecules that are related to these compounds both functionally and biosynthetically; they are fatty acids and their derivatives. [24,25] In general, lipids are insoluble in water and can be recognized by their hydrophilic-lipophilic balance (HLB), melting point, and fatty acid makeup. For prolonged release, lipids having a high melting point and low HLB are appropriate. On the other hand, excipients with high HLB and semi-solids function as enhancers of bioavailability and immediate release, respectively. [26,27] It is found that triglycerides that

solidify at 25°C make good core materials since they reduce the o/w emulsion's permissible storage life. Emulsomes are composed of unbranched fatty acids with chain lengths varying from c-10 to c-18 and are produced using triglycerides.[22]

Negatively charged particles

Emulsomes can be modified with negatively charged lipid particles, like oleic acid, or negatively charged phospholipids such as phosphatidic acid, phosphatidylinositol, and phosphatidylserine. This enhances the zeta potential of the composition and stabilizes the particles. Additionally, the incorporation of these negatively charged lipids promotes the formation of oppositely charged phospholipid bilayers. [22] As a result, the phospholipid bilayers surrounding the lipid core enhance the loading of the aqueous compartment. The aqueous space between the bilayers generates electrostatic repulsion, driving this effect. The negative charge helps prevent particle aggregation, thereby reducing the chances of coalescence, flocculation, or fusion.[28]

Surfactant

It is important to consider the Hydrophilic Lipophilic Balance (HLB) value when selecting a surfactant. Any surfactant's capacity to produce vesicles can be determined by looking at its HLB number; a number between 4 and 8 was shown to be compatible with vesicle production.[29] Additionally, surfactant transition temperature affects drug entrapment in vesicles. The spans with the highest phase transition temperature are where the medication is most confined, and vice versa.[30,31] A high phase transition temperature and low permeability restrict how much medicine can leak out of the vesicles.[32] A high HLB value of Span 40 and 60 results in a decrease in surface free energy, which allows for the creation of larger vesicles and, in turn, a larger area exposed to the dissolution media.[33,34]

Phosphatidylcholine

One of the main ingredients of lecithin is phosphatidyl choline. The water solubility of phosphatidyl choline is poor. Its phospholipids can form lamellar structures, micelles, or bilayer sheets in aqueous solution based on temperature and moisture content. As a result, a kind of surfactant known as amphipathic is produced. They are a vital part of biological membranes and are readily obtained from a variety of readily available sources, including soy beans and egg yolks. Depending on where they come from, they are referred to as egg lecithin or soy lecithin. The drug entrapment rate increased to 96.1% when lecithin was added, and the vesicles contracted as a result of their increased hydrophobicity.[35]

Cholesterol

Emulsomes need cholesterol as a required component because they are vesicles. Vesicle stability is impacted by the addition of cholesterol.[36,37] The amount of cholesterol present is crucial for the drug's trapping in vesicles.[38,39] There have been indications that an increase in cholesterol content increases entrapment efficiency. It was shown that a very high cholesterol content reduced the amount of drugs that were entrapped in the vesicles.[40] This could be as a result of cholesterol starting to interfere with the natural bilayer structure after a certain point, which causes the loss of drug entrapment.[41] Various materials used for a preparation of emulsomes like a Cholesterol, Triglycerides, Soya lecithin, Stearyl amine, Antioxidants, Surfactants. The materials used to prepare emulsomes are given in a [Table.1]

Sr.no	Material	Uses
1	Cholesterol	Incorporating cholesterol affects the stability of vesicles; however, excessive cholesterol can cause the formulation to become unstable.
2	Surfactants	Assign the drug's maximum entrapment.
3	Triglycerides	Low HLB lipids provide a formulation that is sustainably released when used as a hydrophobic lipid core.
4	Antioxidants	Prevent the lipids from becoming rancid or oxidized.
5	Soya lecithin	Additionally, it improves the entrapment efficiency of lamellar structures, micelles, or bilayer sheets.

Table.1 materials used for emulsome preparation with their uses.

Advantages of emulsomes –

- Drug concentration increased in damaged tissue.
- Low systemic absorption.
- Reduced toxicity
- A safe and efficient profile of cytotoxicity.
- Modify the drug's pharmacokinetics.
- Stop the development of resistance to many drugs.
- Slow medication release profile prolongs drug efficacy.
- Exceptional stability
- poorly water-soluble drugs have a high load capacity.
- pharmacological efficacy Improved while lowering toxicity.
- Make medications that are poorly soluble in water more soluble and bioavailable.[42]

Disadvantages of emulsomes -

- In parenteral administration, surfactants are used sparingly because of their detergent-like properties.
- During parenteral administration, it causes adverse effects.
- Drug loading capacity is limited.

- The excessive oil concentration compromises the stability of the formulation.[43]

Methods of preparation



Lipid film formation (Handshaking method)

This method entails casting lipids or surfactants as layers of film that make up their organic solution using a flask rotary evaporator at low pressure (or by hand shaking). After that, an aqueous medium is used to spread the cast films. [44] This procedure calls for constant, mild shaking. It is possible to apply the mechanical energy required to cause the lipid to swell and the cast lipid film to disperse by shaking the film by hand or by exposing it to nitrogen-saturated water steam for fifteen minutes, after which it can swell in the aqueous medium without shaking. While the hand shaking procedure produced multi-lamellar vesicles (MLV), the non-shaking strategy produced large unilamellar vesicles (LMV)

Reserve phase evaporation

The Reserve phase evaporation this technique also called REV method and it is comprised in a two steps. Prepare a phospholipid emulsion in water with an excess of organic phase buffer. With less pressure, remove the organic phase. Usually, mechanical techniques or sonication are used to emulsify the two phases (phospholipids and water). When the organic solvent is removed under vacuum, the water droplets covered with phospholipids clump together to produce a matrix that resembles gel. When the organic solvent is further removed and the pressure is lowered, the gel-like matrix transforms into a smooth paste. This paste is a LUV suspension.[46] This technique can yield drug entrapment efficiencies of 60–65%. Large and small compounds were both encapsulated using this technique.[47] In this process, organic solvents like freon, isopropylether, or chloroform are used to dissolve phospholipids. This procedure calls for constant, mild shaking. It is possible to apply the mechanical energy required to cause the lipid to swell and the cast lipid film to disperse by shaking the film by hand or by exposing it to nitrogen-saturated water steam for fifteen minutes, after which it can swell in the aqueous medium without shaking. While the hand shaking procedure

produced multi-lamellar vesicles (MLV), the non-shaking strategy produced large unilamellar vesicles (LMV).(figure-4)

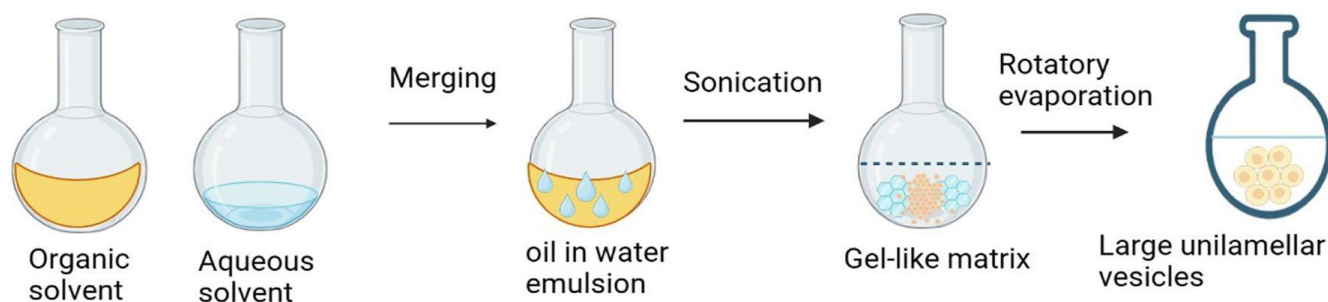


Figure 4: Reserve phase evaporation

Detergent removal technique

This method creates micellar mixtures by combining or mixing phospholipids with a detergent. The detergent was then eliminated from the mixture as the phospholipid content of the micelles increased and, at last, the lipids united to form single bilayer vesicles. These methods, which include adsorption onto bio beads, dialysis, and column chromatography, can remove the detergent from the formulation.[48] Additionally, emulsome preparation can be done using this technique. Typically, high critical micelle concentration (CMC) detergents are utilized for this purpose. Materials with high CMC (between 10 and 20 mM) such as octylglycoside, sodium cholate, sodium deoxycholate, and other detergents are suitable for this purpose.[49] In this method, the phospholipid detergent combination was sent through a dialysis cell to remove the detergent. This method was claimed to create a homogeneous population of single-layered emulsomes with mean sizes between 50 and 100 nm. [50,51] (figure-5)

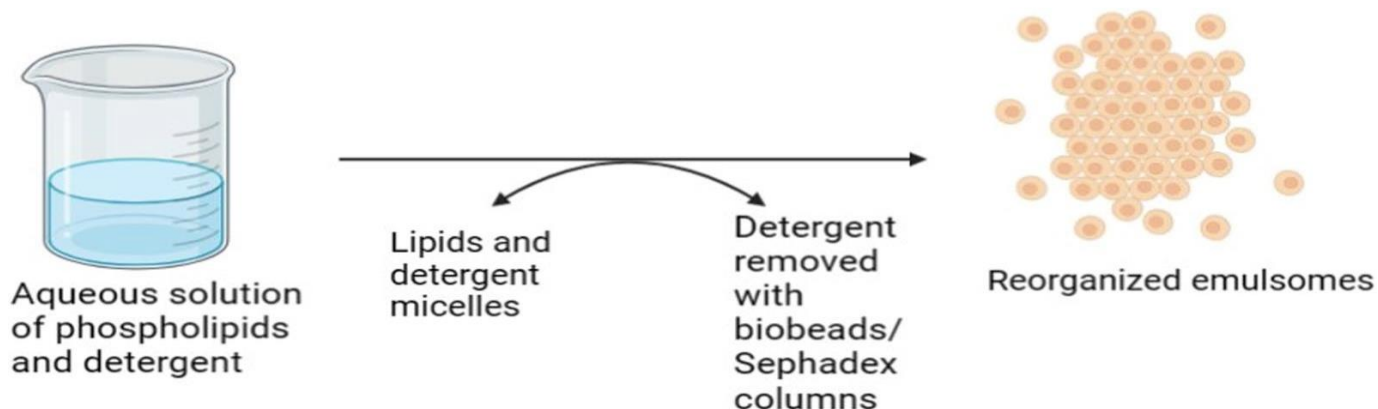


Figure 5: Detergent removal technique

Ethanol injection method

This method is one of the alternatives used to produce small unilammellar vesicles (SUVs), according to documentation. Using a small needle, a surfactant solution composed of ethanol is rapidly injected into an excess of saline or another aqueous media.[52] Ethanol forms vesicles when it evaporates. A tight distribution of tiny liposomes (less than 100 nm) can be created by the ethanol injection method, which involves merely injecting an ethanolic lipid solution in water or in a single step without extrusion or sonication. The right ethanol injection strategy can generate emulsomes with a moderate average radius on their own. [53,54] (figure-6)

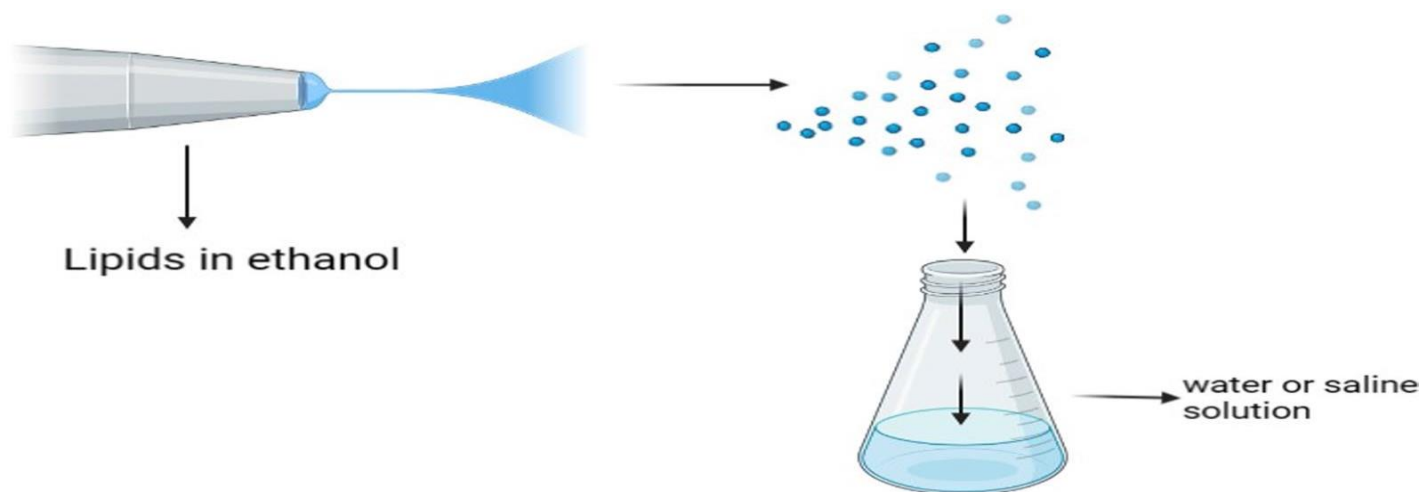


Figure 6 : Ethanol injection method

High-pressure extrusion technique

Numerous studies have demonstrated that when MLV are repeatedly transferred under high pressure through very small pore polycarbonate membranes (0.8 to 1.0 μm), the average diameter of the vesicles progressively shrinks, reaching a minimum of 60 to 80 nm after five to ten passes. When vesicles lose their typical size, they tend to become unilamellar.[55,56,57] Similar results have been reported by other studies when MLV is passed through a microfluidizer. A microfluidizer is a machine that forces feed material through a tiny hole using high pressure. Similar to how onion skin layers separate during peeling, it appears that when MLV are forced through the microscopic opening, bilayer layers are taken from the vesicular structure. Furthermore, it was suggested that the layer separation mechanism only applied to vesicles made of positively charged phospholipids and larger than 70 μm .[58] (figure-7)

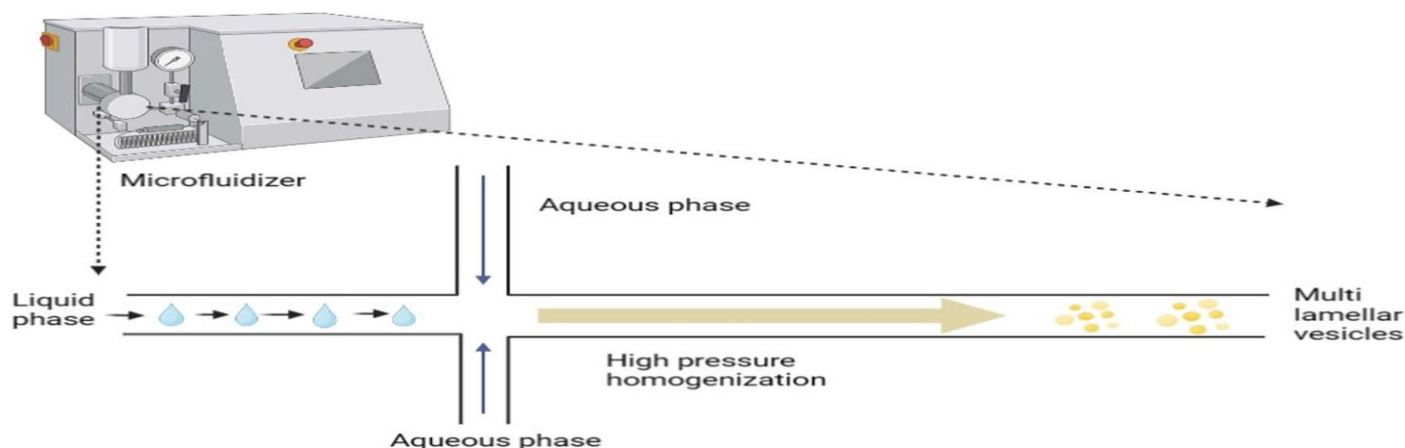


Figure 7 : High-pressure extrusion technique

Cast film method

When combined with phospholipids at a weight ratio of 0.5:1.0, triglycerides can undergo a phase transition from solid to liquid at temperatures over 25°C. The combination must be suspended in an aqueous solution at a temperature lower than the solid-to-liquid transition temperature in order to produce emulsomes. The liquid particle nano-emulsion that makes up these emulsomes has a mean diameter of 10–250 nm and a frequency of 50–150 nm. Usually, the particles fall between 20 and 180 nm in size. It is preferable to define the size range based on weight percentage as opposed to particle number. The lipid component may typically be a volatile organic solvent that is chemically non-reactive, such as dichloromethane or diethylether. A low-pressure rotary evaporator or an inert gas stream are typically used to remove the solvent. To hydrate and disseminate the resulting lipid layer, cover and shake with an aqueous solution. If the drug component wasn't in the organic solution, it may be added to the aqueous hydration solution. After that, the lipid suspension or dispersion is typically sized at pressures of up to 800 bar using a high shear homogenizant.[19]

Sonication method

This procedure involved dissolving solid lipids, cholesterol, and phosphatidylcholine in a little amount of chloroform that contained three or four drops of methanol in a round-bottom flask. The additions were made in different molar ratios. In this solution, a precisely weighed quantity of drug was dissolved. Using a rotary evaporator, the organic solvent was evaporated until it was entirely dry at reduced pressure, forming a thin lipid layer on the walls of the round bottom flask. [59,60,61] In order to produce nanoscale emulsomes, 10 milliliters of phosphate-buffered saline (pH 7.4) were used to hydrate the dry film. After that, ultrasonication was used for 15 minutes at 40% frequency to homogenize it.[62,63] (figure-8)

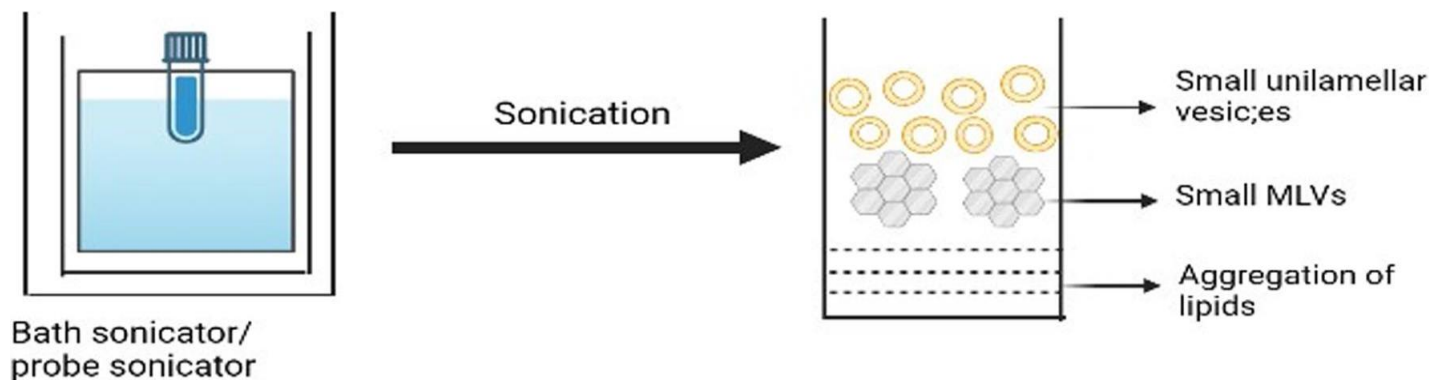


Figure 8: Sonication method

Evaluation of emulsomes –

1) Entrapment efficiency (EE%)

Centrifugation was used for 1 hour at 15,000 rpm at 4°C to separate the non-entrapped medication from the emulsomes in the cooled samples.[64] Using 1% Triton X to dissolve emulsomes and spectrophotometrically detecting the drug's concentration at λ_{max} 305 nm, the EE% of oxcarbazepine was ascertained.[65] The following formula was used to determine the EE%

$$EE\% = \frac{\text{Entrapped drug}}{\text{Total drug}} \times 100$$

2) Transmission electron microscopy (TEM)

A drop of 1% phosphotungstic acid was then added to the emulsomes to negatively stain them. A 50 mL drop of the vesicular dispersion was put on a carbon-coated grid, left to adsorb on the carbon film for two minutes, and any extra liquid was wiped out with fiber-free, toughened filter paper to image a sample of the generated emulsomal particles. The grid was cleaned with distilled water, any leftover discoloration was removed with filter paper, and it was then allowed to air dry. The sample underwent a 200 kV TEM analysis with a magnification of X 25000.[66]

3) Drug release

OX-bearing emulsomes' permeability through the cellophane membrane was investigated in Franz-type diffusion cells. Three milligrams of medication's worth of emulsomes were present in the donor compartment. The receptor compartment, which was maintained at 37 ± 0.2 °C by a flowing water jacket, contained 7.5 mL of phosphate buffer (pH 6.8) that was continuously mixed by a small magnetic bar rotating at 50 rpm.[67] In order to replicate the low mixing conditions found in the nose, this low stirring rate was used.[68]

4) Particle characterization

The freshly manufactured emulsomes' average particle size and size distribution were estimated using dynamic light scattering (DLS) and represented by the polydispersity index (PDI). To find the zeta potential, or ζ , electrophoretic light scattering was employed with a Zetasizer Nano ZS fitted with 4 mW 633 nm He-

Ne lasers. Samples were allowed to equilibrate at 25°C after being suitably diluted with deionized water to avoid multi-scattering. Analysis was conducted at an angle of 173 degrees.[69]

Stability of emulsomes

The ability of the nanocarrier to sustain its biophysical properties over time, such as size, zeta potential, and drug retention, is known as stability in relation to DDSs mediated by nanoparticles. Because of their high absolute levels of zeta potential, emulsomes are consequently predicted to be physically stable, which lowers the likelihood of coalescence. [70] Emulsomes' superior stability in suspensions over liposomes and other lipid-based formulations may prove to be very advantageous in therapeutic settings.[71] Two essential components are combined to create emulsomes: phospholipid, which gives the lipidic core its vesicular steric stability by enclosing it. It is feasible to create a pharmaceutically stable emulsomal formulation without the need for an extra surfactant or solubilizer. PEGylation of the emulsome surface will therefore help to increase steric stability and extend circulation time.[72,73]

The stability of the nanocarrier is significantly influenced by the physicochemical characteristics of the lipids used and the storage temperature.[74] Zeta potential is a crucial component of stability, and in order to study the storage stability of electrostatically stabilized vesicles and comprehend the workings of dispersion and aggregation processes, it is unquestionably vital to measure the zeta potential in these containers. When the kind of lipid was altered, the zeta potential improved, but it was shown to be notably negative for the fats used in emulsomes, such as tristearin, compritol ATO 888, and trilaurin. More energy is added during sonication, resulting in a decrease in particle size and zeta potential and the production of stable, tightly packed emulsomes. [72,75]

Biopharmaceutical aspects

The solubility of inefficient luminal medicines is increased with the help of emulsomes. The GI tract's lipids increase the secretion of endogenous biliary lipids, including phospholipid and cholesterol. This leads to the formation of intestinal mixed micelles that contain phospholipid, cholesterol, and bile salts and improves the GI tract's ability to solubilize substances. But when exogenous lipids are added to these bile salt complexes, either directly or indirectly through digestion, the micellar structures grow and can become more soluble. [76] When intestinal lymphatic transport is stimulated, lipids may either directly or indirectly increase bioavailability by decreasing first-pass metabolism or by increasing lymphatic transport.

Emulsomes have the potential to improve bioavailability by altering the GI tract's biochemical barrier function. As demonstrated by the p glycoprotein efflux pump, it is evident that the lipids and triglycerides included in the emulsomal preparation may lessen the amount of enterocyte-based metabolism as well as the activity of intestinal efflux transporters.[77,78]

The physical barrier function of the GI tract may be changed by emulsomes. Certain mixtures of lipids and triglycerides have been shown to have properties that increase permeability. [79,80]

Application of Emulsomes

Drug targeting

The ability of emulsomes to target medications is one of their greatest benefits. The reticulo-endothelial system is a target for medication delivery via emulsomes. Emulsome absorption is favored by the reticuloendothelial system. Circulating blood factors known as opsonins regulate the consumption of Emulsomal vesicles. To be cleared, these opsonins identify the vesicles. Drugs with this specific location are used to treat malignancies in animals whose liver and spleen are known to metastasis. Hepatic parasite infections can also be treated with this drug fixing method. In addition to the reticuloendothelial system, emulsomes can be utilized to direct medications to other organs. Vesicles have the ability to be targeted to particular organs by attaching a carrier system to them.[81,82]

Increase bioavailability of lipophilic drugs

Emulsomes have been shown to be an effective drug delivery vehicle for lipophilic medicines, which have low water solubility in biological fluids and hence low drug absorption and bioavailability. Emulsomes are made of solid lipid at their center, which entraps lipophilic medicines and releases them gradually. Emulsomes also benefit from having a higher drug content, which lowers dosage frequency and improves entrapment efficiency.[83]

Anti-neoplastic treatment

The adverse effects of most antineoplastic drugs are immediate. By altering a drug's metabolism, lengthening its half-life, and boosting its circulation, emulsomes might lessen the negative effects of the medication. When compared to the untrapped medicines, methotrexate exhibited favorable effects from enzymatic entrapment, including a slowed rate of tumor development and increased plasma levels that were followed by a delayed clearance.[81,84]

Leishmaniasis

Leishmaniasis is caused by a parasite from the genus *Leishmania* that infects the cells of the liver and spleen. Antimony compounds, which can harm the kidneys, liver, and heart in excessive doses, are commonly prescribed medications for the condition. Emulsome was used in experiments to demonstrate that bigger dosages of the medication may be given without causing side effects, increasing treatment efficacy. [81,85]

AIDS treatment

The medication zidovudine is approved to treat AIDS. Due to its high lipophilicity, the medication has altered pharmacokinetics and some major adverse effects. Low bioavailability of zidovudine and other side effects have been resolved by encapsulating the drug in emulsomes.

Dermal therapy

Dithranol has been used to treat psoriasis, yet the medication has unfavorable side effects that include erythema, peeling, staining, and skin irritation. However, the addition of dithranol to emulsomes improves medication retention in skin tissue and increases skin penetration while reducing negative effects. [86]

Future Aspects

Emulsomes represent a potentially effective method for the formulation of pharmaceutical substances exhibiting variable oral bioavailability and poor aqueous solubility. Emulsomes can facilitate the oral delivery of hydrophobic medicines, since they have demonstrated a significant improvement in oral bioavailability. Emulsome renaissances throughout the last few decades are drawing more and more attention. The last few decades are drawing more and more interest. An increase in oral bioavailability cannot be predicted using many in vitro models, and most emulsome development is still empirical. The creation of in vitro methods for predicting the dynamic changes involving the drug in the gut is necessary for tracking the medication's solubilization condition in vivo. The interactions between the pharmacologically active ingredient and lipid systems also require consideration. In order to create recommendations that enable the early identification of viable candidate formulations, it is also necessary to comprehend the characteristics of diverse lipid formulations. Both further basic research on the mechanisms of action of this fascinating and diverse collection of formulations and future studies or research on human bioavailability are required.

Conclusion:

Emulsomes' distinct structural makeup and adaptable properties make them a promising development in the realm of medication delivery systems. Emulsomes combine the advantages of liposomes with emulsions to provide sustained release profiles, increased stability, and better drug loading capacity. They are extremely

versatile for a range of therapeutic applications, such as controlled release systems and targeted drug administration, because they can encapsulate both hydrophilic and hydrophobic medications. Additionally, site-specific targeting can be accomplished by functionalizing emulsomes, increasing therapeutic efficacy while reducing adverse effects. Even with these encouraging characteristics, more research is required to improve their formulation, look into long-term stability, and assess their efficacy and safety in clinical settings. In conclusion, emulsomes have a lot of promise as a new vesicular platform for targeted therapy and improved drug delivery, opening the door for more developments in pharmaceutical formulations in the future.

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